

इंटरनेट

मानक

Disclosure to Promote the Right To Information

Whereas the Parliament of India has set out to provide a practical regime of right to information for citizens to secure access to information under the control of public authorities, in order to promote transparency and accountability in the working of every public authority, and whereas the attached publication of the Bureau of Indian Standards is of particular interest to the public, particularly disadvantaged communities and those engaged in the pursuit of education and knowledge, the attached public safety standard is made available to promote the timely dissemination of this information in an accurate manner to the public.

“जानने का अधिकार, जीने का अधिकार”

Mazdoor Kisan Shakti Sangathan

“The Right to Information, The Right to Live”

“पुराने को छोड़ नये के तरफ”

Jawaharlal Nehru

“Step Out From the Old to the New”

IS 9873-3 (1999): Safety Requirements for Toys, Part 3:
Migration of Certain Elements [PCD 12: Plastics]



“ज्ञान से एक नये भारत का निर्माण”

Satyanarayan Gangaram Pitroda

“Invent a New India Using Knowledge”



“ज्ञान एक ऐसा खजाना है जो कभी चुराया नहीं जा सकता है”

Bhartrhari—Nitiśatakam

“Knowledge is such a treasure which cannot be stolen”

BLANK PAGE



भारतीय मानक
खिलौनों की सुरक्षा हेतु अपेक्षाएँ
भाग 3 कुछ तत्वों का स्थानान्तरण
(पहला पुनरीक्षण)

Indian Standard
SAFETY REQUIREMENTS FOR TOYS
PART 3 MIGRATION OF CERTAIN ELEMENTS
(*First Revision*)

ICS 97.200.50

© BIS 1999

BUREAU OF INDIAN STANDARDS
MANAK BHAVAN, 9 BAHADUR SHAH ZAFAR MARG
NEW DELHI 110002

NATIONAL FOREWORD

This Indian Standard (Part 3) (First Revision) which is identical with ISO 8124-3 : 1997 'Safety of toys — Part 3 : Migration of certain elements' issued by the International Organization for Standardization (ISO) was adopted by the Bureau of Indian Standards on the recommendation of Plastics Sectional Committee and approval of the Petroleum, Coal and Related Products Division Council.

This standard was first published in 1981 which was based on BS 5663 : Part 2 : 1978 (EN 71 : Part 2) 'Safety of toys, Part 2 Flammability of toys', issued by the British Standards Institution. The Committee decided to revise this standard with the adoption of ISO 8124-3 : 1997 under dual numbering system. Consequently, the title has been modified as 'Safety requirements for toys : Part 3 Migration of certain elements'.

The text of ISO standard has been approved as suitable for publication as Indian Standard without deviations. Certain conventions are, however, not identical to those used in Indian Standards. Attention is particularly drawn to the following:

- a) Wherever the words 'International Standard' appear referring to this standard, they should be read as 'Indian Standard'.
- b) Comma (,) has been used as a decimal marker while in Indian Standards, the current practice is to use a point (.) as the decimal marker.

In this adopted standard, reference appears to certain International Standards for which Indian Standards also exist. The corresponding Indian Standards which are to be substituted in their place are listed below along with their degree of equivalence for the editions indicated:

<i>International Standard</i>	<i>Corresponding Indian Standard</i>	<i>Degree of Equivalence</i>
ISO 8124-1 Safety of toys — Part 1 : Mechanical and physical properties	IS 9873 (Part 1) : 1981 Safety requirements for toys : Part 1 Mechanical and physical properties	Not technically equivalent. However this Indian standard will be considered for revision to align it with ISO 8124-1 when the same is published
ISO 3696 : 1987 Water for analytical laboratory use — Specification and test methods	IS 1070 : 1992 Reagent grade water (<i>third-revision</i>)	Technically equivalent

In reporting the results of a test or analysis made in accordance with this standard, if the final value, observed or calculated, is to be rounded off, it shall be done in accordance with IS 2 : 1960 'Rules for rounding off numerical values (*revised*)'.

AMENDMENT NO. 3 JULY 2011
TO
IS 9873 (PART 3) : 1999/ISO 8124-3 : 1997
SAFETY REQUIREMENTS FOR TOYS
PART 3 MIGRATION OF CERTAIN ELEMENTS

(First Revision)

(First cover, pages 1 to 21, Designation) — Substitute ‘IS 9873 (Part 3) : 1999’ for ‘IS 9873 (Part 3) : 1999/ISO 8124-3 : 1997’ wherever it appears in the standard.

(First cover and page 1, Title) — Substitute the following for the existing title:

SAFETY REQUIREMENTS FOR TOYS
PART 3 MIGRATION OF CERTAIN ELEMENTS AND PHTHALATES

(First Revision)

[Second cover, National Foreword, Para 1 (see also Amendment No. 2)] — Substitute the following two paras for the existing para 1:

‘FOREWORD

This Indian Standard (Part 3) (First Revision), a modified adoption of International Standard ‘ISO 8124-3 : 1997 ‘Safety requirements for toys — Part 3: Migration of certain elements’, was adopted by the Bureau of Indian Standards on the recommendation of Plastics Sectional Committee and approval of the Petroleum, Coal and Related Products Division Council. This standard was initially published in 1999 as identical to ISO 8124-3 : 1997, issued by International Organization for Standardization. Since the requirement of phthalates is yet to be incorporated in ISO standard, the dual number status of this standard has been changed to modified adoption for incorporating the safe limits and test method for certain phthalates.

Phthalates are used as plasticizers (softening agents) in the manufacture of soft vinyl [also known as polyvinyl chloride (PVC)], which in turn is used in toys and childcare articles. Phthalates do not bind to the soft vinyl, but are present as mobile components of the vinyl. However, the mere presence of phthalates in soft vinyl does not in itself equate to a health risk. It is the amount of phthalates that leach out of soft vinyl and are absorbed into the body that can be harmful. Although exposure to phthalates in soft vinyl through dermal contact or inhalation

Price Group 2

Amend No. 3 to IS 9873 (Part 3) : 1999/ISO 8124-3 : 1997

is negligible, phthalates may leach out of soft vinyl during periods of sustained mouthing action (sucking and chewing, but not licking) and enter the body through the saliva. Once in the body, some phthalates have the potential to cause adverse effects on reproduction and development. It is for this reason that a need was felt to specify the safe limits of phthalates in toys through amendment.

(Page 1, clause 1) — Substitute the following for the existing:

‘This part of Indian Standard specifies maximum acceptable levels and method of sampling and extraction prior to analysis for migration of elements antimony, arsenic, barium, cadmium, chromium, lead, mercury and selenium and phthalates from toys materials and parts of toys, except material not accessible (see ISO 8124-1).’

(Page 12, clause 10) — Insert the following new clause at the end:

11 Maximum acceptable levels of phthalates

When tested in accordance with the method given at clause 11.1, the toy shall not show the presence of phthalates beyond the limits given as under:

Characteristic	Requirement
Vinyl in toys or childcare article	≤ 0.1% of: i) Bis (2-ethyl hexyl) phthalate (DEHP) ii) Dibutyl phthalate (DBP) or iii) Benzyl butyl phthalate (BBP)
Vinyl in any part of the toy or childcare article that can be placed in mouth of a child under 4 years of age	≤ 0.1% of: i) Di-isononyl phthalate (DINP) ii) Di-isodecyl phthalate (DIDP) or iii) Di- <i>n</i> -octyl phthalate (DNOP)

11.1 Method of test for determination of phthalates in toys

11.1.1 Principle

The method is used for quantitative determination of the following six phthalates in toys made of Poly Vinyl Chloride.

- a) Bis (2-ethyl hexyl) phthalate (DEHP)
- b) Dibutyl phthalate (DBP)

- c) Benzyl butyl phthalate (BBP)
- d) Di-isononylphthalate (DINP)
- e) Di-isodecyl phthalate (DIDP)
- f) Di-*n*-octyl phthalate (DNOP)

The method involves extraction of these phthalates using dichloromethane and quantification of individual phthalates through Gas Chromatography – Mass Spectrometric technique.

11.1.2 Apparatus

11.1.2.1 Gas Chromatograph with Mass Spectrometer (GC-MS) with all required accessories including syringes, analytical columns with appropriate data system. The suggested operative parameters for GC-MS are as given below, but can be changed, provided standardization is done.

11.1.2.1.1 GC conditions

Column: 30 m × 0.25 mm × 0.25 μm (5% phenyl-methyl polysiloxane)
Inlet Temperature: 280°C
Injection Volume: 1 μl
Carrier Gas Flow: Helium at 1ml/min
Injection Mode: Splitless, pulse injection at 35 psi for 0.6 min
Oven Program: 50°C for 2 min, to 280°C at 30°C/min, to 310°C at 15°C/min
and hold for 3.33 min, for a total run time of 15 min
Washing Solvent: Hexane

11.1.2.1.2 MS conditions

Filament Delay: 4 min
Ion Trap Temp: 220°C
Manifold Temp: 50°C
Transfer Line Temp: 280°C

11.1.2.2 Soxhlet assembly, comprising of Soxhlet apparatus, round bottom flask of suitable capacity (say 150 ml, 250 ml or 500 ml), thimble, water cooled condenser, heating mantle.

11.1.2.3 Balance, capable of weighing accurately upto 0.1 mg.

11.1.2.4 Volumetric flasks, of suitable capacities.

Amend No. 3 to IS 9873 (Part 3) : 1999/ISO 8124-3 : 1997

11.1.3 Reagents

11.1.3.1 Certified reference material of following six phthalates

Compound	CAS No.
Bis (2-ethyl hexyl) phthalate (DEHP)	117-81-7
Dibutyl phthalate (DBP)	84-74-2
Benzyl butyl phthalate (BBP)	85-68-7
Di- <i>n</i> -octyl phthalate (DNOP)	117-84-0
Di-isononyl phthalate (DINP)	28553-12-0
Di-isodecyl phthalate (DIDP)	26761-40-0

11.1.3.2 Dichloromethane (*spectroscopy grade or higher*)

11.1.3.3 Hexane (*spectroscopy grade or higher*)

11.1.3.4 Internal Standard: Benzyl Benzoate (CAS No. 120-51-4)

11.1.4 Procedure

11.1.4.1 Preparation of standard solutions

- a) High Concentration Stock Solution (S1) — Weigh accurately 0.1 g each of DEHP, DBP, BBP, DNOP and 0.4 g each of DINP and DIDP (accurately upto 0.1 mg) individually in separate 100 ml volumetric flasks. Add Hexane to the mark to make 1 000 µg/ml concentration stock solution for DEHP, DBP, BBP & DNOP and 4 000 µg/ml concentration stock solution for DINP and DIDP.
- b) Middle Concentration Stock Solution (S2) — Add 1 ml each of DEHP, DBP, BBP, DNOP, DINP and DIDP to a 10 ml volumetric flask. Add Hexane to the mark to make the Middle Concentration Stock Solution (S2) containing 100 µg/ml each of DEHP, DBP, BBP and DNOP and 400 µg/ml each of DINP and DIDP.
- c) Internal Standard (IS) — Weigh accurately 0.1 g of Benzyl Benzoate (BB) in a 100 ml volumetric flask and make up the volume with Hexane to make stock solution of 1 000 µg/ml concentration of Benzyl Benzoate. Add 1 000 µl of the stock solution to a 10 ml volumetric flask to make Benzyl Benzoate Internal Standard (IS) of 100 µg/ml for

Amend No. 3 to IS 9873 (Part 3) : 1999/ISO 8124-3 : 1997

use in preparation of calibration curve. Add 500 µl of the stock solution to another 10 ml volumetric flask to make 50 µg/ml concentration of Benzyl Benzoate for use in sample testing.

11.1.4.2 Preparation of calibration curves — Prepare calibration curves using standard solutions ranging from 0.5 to 5 µg/ml (or any other suitable range). A suggested plan for preparing the calibration curve is given under Table 1.

Table 1 Plan for Preparation of Calibration Curves

Calibration Solution	Volume in µl of S2	Volume in µl of Internal Standard of 100 µg/ml Conc	Final Volume (ml)	Concentration (µg/ml)	
				DEHP, DBP, BBP, DNOP	DINP, DIDP
Level 1	50	100	10	0.5	2
Level 2	100	100	10	1	4
Level 3	200	100	10	2	8
Level 4	300	100	10	3	12
Level 5	400	100	10	4	16
Level 6	500	100	10	5	20

NOTE — Each calibration Standard to have an internal standard concentration of 1 µg/ml.

11.1.4.3 Sample preparation

- i) Select only PVC part of the toys by removing metal parts, ornaments, clothes, hair, etc, if present with the toy.
- ii) Cut or chop or grind the PVC part of the toy sample to fine pieces/powder. Particle size upto 0.5 mm is desirable.
- iii) In case, the grind/cut material is required to be stored, it may be stored preferably in non-plastic material, preferably glass material.

11.1.4.4 Extraction and estimation

- a) Place 0.5 g to 1 g of material accurately weighed in a paper thimble.
- b) Place the thimble into the Soxhlet Apparatus.
- c) Add dichloromethane to the round bottom flask upto 80% minimum of its full capacity.

Amend No. 3 to IS 9873 (Part 3) : 1999/ISO 8124-3 : 1997

- d) Start Soxhlet extraction for about 2 h ensuring uniform rate of extraction. A rate of 10 cycles/h is considered suitable for this purpose.
- e) On completion of the extraction, the extract be allowed to cool down to room temperature.
- f) Remove the solvent from the extract by using suitable evaporator (rotary, turbo etc.) under suitable conditions depending upon the evaporator.
- g) Dissolve the residue with Hexane and transfer the solution to a 10 ml volumetric flask by repeatedly rinsing the flask with Hexane for ensuring complete transfer.
- h) Make up the volume upto the mark (10 ml) with Hexane.
- j) Combine 100 μ l* of the solution with 20 μ l of internal standard (BB, 50 μ g/ml) and 880 μ l of hexane in a GC-MS vial. Mix thoroughly and analyze with GC-MS.

*Depending on the phthalate content in toy samples, a different dilution ratio may be necessary to bring the concentration levels within the calibration range.

11.1.4.5 Illustrative chromatograms — A typical Total Ion Chromatogram for the six phthalates is shown in Fig. 1. As could be seen, four phthalates, namely, DBP, BBP, DEHP and DNOP appear as sharp single peaks whereas the remaining two phthalates, namely, DINP and DIDP appear as hump (or finger peaks).

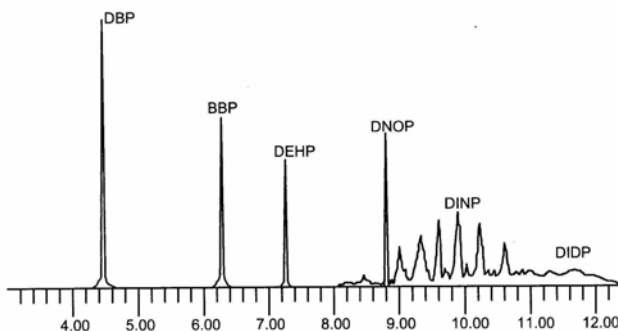


FIG.1 TYPICAL TOTAL ION CHROMATOGRAM FOR DEHP, DBP, BBP, DINP, DIDP AND DNOP

Amend No. 3 to IS 9873 (Part 3) : 1999/ISO 8124-3 : 1997

The finger peaks could appear in the Total Ion Chromatograms for DINP and DIDP due to possibilities of number of isomers. It is recommended to use area summation integration technique for quantification of these phthalates. Typical Extracted Ion Chromatogram for these two phthalates are given in Fig. 2 for guidance:

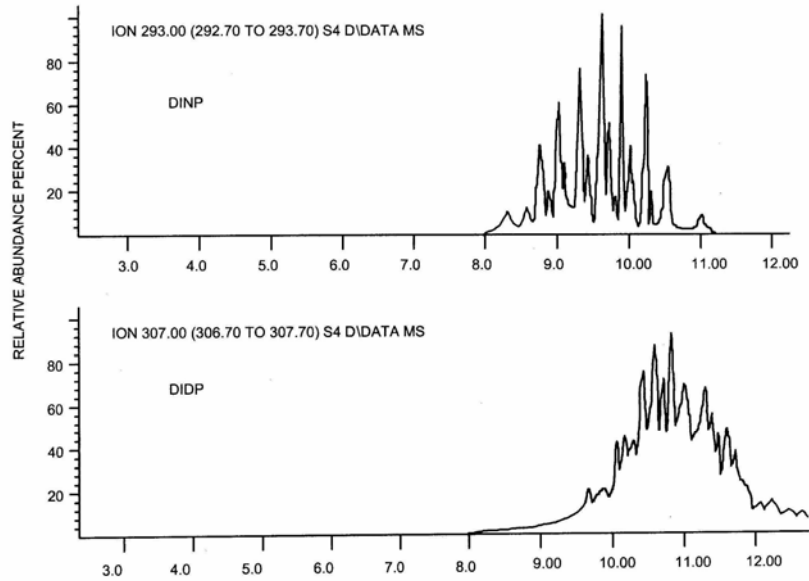


FIG. 2 TYPICAL EXTRACTED ION CHROMATOGRAM FOR DINP AND DIDP

11.1.5 Results

11.1.5.1 Qualitative Analysis — Identification of phthalates present in the toy samples be done by comparing the retention time of sample peaks with peaks of standard compounds which shall be run with the same equipment under similar conditions. Typical quantification ions for the six phthalates are given under Table 2. Each ion produced from the same compound must have maximum intensity at the same scan or plus/minus one scan.

Table 2 Typical Quantification Ions

Phthalate	Characteristic Ions (m/z)			
	Target ion (m/z)	Qualification Ion 1 (m/z)	Qualification Ion 2 (m/z)	Qualification Ion 3 (m/z)
DBP	149	150	223	205
BBP	149	91	206	238
DEHP	149	167	279	150
DNOP	279	149	150	261
DINP	293	149	127	167
DIDP	307	149	141	150

11.1.5.2 Quantitative analysis

Calculate the concentration of phthalate in sample using the following formula:

$$\text{Phthalate content, percent by mass} = \frac{A \times V}{M} \times F \times (1 \times 10^{-6}) \times 100$$

where

- A = concentration in extract, $\mu\text{g/ml}$;
- V = final volume of extract in ml;
- M = mass of sample in g; and
- F = dilution factor.

AMENDMENT NO. 2 OCTOBER 2010
TO
IS 9873 (PART 3): 1999/ISO 8124-3 : 1997 SAFETY
REQUIREMENTS FOR TOYS

PART 3 MIGRATION OF CERTAIN ELEMENTS

(First Revision)

(Amendment No. 1) — Delete.

(National Foreword, last paragraph) — Insert the following new paragraph at the end:

'BIS Certification Marking Clause is given in National Annex A.'

(Page 22) — Insert the following National Annex A at the end of the standard:

NATIONAL ANNEX A
(National Foreword)

A-1 BIS CERTIFICATION MARKING

The product may also be marked with the Standard Mark.

A-1.1 The use of the Standard Mark is governed by the provisions of the *Bureau of Indian Standards Act, 1986* and the Rules and Regulations made thereunder. The details of conditions under which the license for use of Standard Mark may be granted to manufacturers or producers may be obtained from the Bureau of Indian Standards.

A-2 MARKING

A-2.1 The following shall be marked legibly and indelibly on each toy:

Name of Manufacturer and/or his authorized representative and/or registered trade mark, if any.

Amend No. 2 to IS 9873 (Part 3): 1999/ISO 8124-3 :1997

A-2.2 The following shall be marked legibly and indelibly on toy packing:

- a) Name of the Article;
- b) Country of origin;
- c) Manufacturer's **or** his authorized representative's name & address and trade mark, if any;
- d) Batch No./Code No.;
- e) Month and year of manufacture;
- f) Instructions, for use and storage as applicable (on packing/leaflet/tag, etc);
- g) Safety directions, if any;
- h) Other provisions of Regulatory Authority, as required; and
- j) Usage's Age Group.

**AMENDMENT NO. 1 AUGUST 2009
TO
IS 9873 (PART 3) : 1999/ISO 8124-3 : 1997 SAFETY REQUIREMENTS FOR TOYS**

PART 3 MIGRATION OF CERTAIN ELEMENTS

(First Revision)

(National Foreword, last paragraph) — Insert the following new paragraph at the end:

“The Standard also makes reference to the BIS Certification Marking of the product, details of which is given in National Annex A.

For the purpose of deciding whether a particular requirement of this standard is complied with, the final value, observed or calculated, expressing the result of a test or analysis, shall be rounded off in accordance with IS 2 : 1960 ‘Rules for rounding off numerical values (*revised*)’. The number of significant places retained in the rounded off value should be the same as that of the specified value in this standard.”

(Page 22) — Insert the following National Annex A at the end of the standard:

NATIONAL ANNEX A
(National Foreword)

A-1 BIS CERTIFICATION MARKING

The product may also be marked with the Standard Mark.

A-1.1 The use of the Standard Mark is governed by the provisions of the *Bureau of Indian Standards Act, 1986* and the Rules and Regulations made thereunder. The details of conditions under which the license for use of Standard Mark may be granted to manufacturers or producers may be obtained from the Bureau of Indian Standards.

A-2 MARKING

A-2.1 The following shall be marked permanently on each article as well as on

Amend No. 1 to IS 9873 (Part 3) : 1999/ISO 8124-3 : 1997

- a) Name of the Article;
- b) Country of origin;
- c) Manufacturer’s or his authorized representative’s name and address, and trade mark, if any;
- d) Batch No./Code No.;
- e) Month and year of manufacture;
- f) Instructions for use and storage;
- g) Safety directions, if any;
- h) Other provisions of Packaging Commodity Act;
- j) Usage’s Age Group; and
- k) Any loose part along with the article or any other information.

Indian Standard
SAFETY REQUIREMENTS FOR TOYS
PART 3 MIGRATION OF CERTAIN ELEMENTS
(First Revision)

1 Scope

1.1 This part of ISO 8124 specifies maximum acceptable levels and methods of sampling and extraction prior to analysis for the migration of the elements antimony, arsenic, barium, cadmium, chromium, lead, mercury and selenium from toy materials and from parts of toys, except materials not accessible (see ISO 8124-1).

1.2 Maximum acceptable levels are specified for the migration of the elements listed in 1.1 from the following toy materials:

- coatings of paints, varnishes, lacquers, printing inks, polymers and similar coatings (see 8.1);
- polymeric and similar materials, including laminates, whether textile-reinforced or not, but excluding other textiles (see 8.2);
- paper and paper board, up to a maximum mass per unit area of 400 g/m² (see 8.3);
- natural or synthetic textiles (see 8.4);
- glass/ceramic/metallic materials, excepting lead solder when used for electrical connections (see 8.5);
- other materials, whether mass-coloured or not (e.g. wood, fibreboard, hardboard, bone and leather) (see 8.6);
- materials intended to leave a trace (e.g. the graphite materials in pencils and liquid ink in pens) (see 8.7);
- pliable modelling materials, including modelling clays, and gels (see 8.8);
- paints to be used as such in the toy, including finger paints, varnishes, lacquers, glazing powders and similar materials in solid or liquid form (see 8.9).

1.3 For the purposes of this part of ISO 8124, the following criteria are considered appropriate in the categorization of toys which can be sucked, licked or swallowed:

- all intended food/oral contact toys, cosmetic toys and writing instruments categorized as toys;
- toys intended for children up to six years of age, i.e. all accessible parts and components where there is a probability that those parts or components may come into contact with the mouth (see annex D).

Toys and parts of toys which, due to their accessibility, function, mass, size or other characteristics, obviously exclude any hazard due to sucking, licking or swallowing, bearing in mind the normal and foreseeable behaviour of children, are not covered by this part of ISO 8124.

1.4 Packaging materials are not included unless they are part of the toy or have intended play value (see annex D).

2 Normative references

The following standards contain provisions which, through reference in this text, constitute provisions of this part of ISO 8124. At the time of publication, the editions indicated were valid. All standards are subject to revision, and parties to agreements based on this part of ISO 8124 are encouraged to investigate the possibility of applying the most recent editions of the standards indicated below. Members of IEC and ISO maintain registers of currently valid International Standards.

ISO 8124-1:—¹⁾, *Safety of toys — Part 1: Mechanical and physical properties.*

ISO 3696:1987, *Water for analytical laboratory use — Specification and test methods.*

3 Definitions

For the purposes of this part of ISO 8124 the following definitions apply.

3.1 base material: Material upon which coatings may be formed or deposited.

3.2 coating: All layers of material formed or deposited on the base material of a toy, including paints, varnishes, lacquers, inks, polymers or other substances of a similar nature, whether they contain metallic particles or not, no matter how they have been applied to the toy, and which can be removed by scraping with a sharp blade.

3.3 detection limit of a method: Three times the standard deviation of the result obtained in the blank test using that method.

3.4 mass-coloured materials: Materials, such as wood, leather and other porous substances, which have absorbed colouring matter without formation of a coating.

3.5 scraping: Mechanical process for removal of coatings down to the base material.

3.6 toy material: All accessible materials present in a toy.

4 Maximum acceptable levels

4.1 Specific requirements

Toys and parts of toys as specified in clause 1 are deemed to meet the requirements of this part of ISO 8124 when the adjusted value of migration of elements from them comply with the maximum limits given in table 1 when tested in accordance with clauses 7, 8 and 9 (see annex D).

4.2 Interpretation of results

Due to the precision of the methods specified in this part of ISO 8124, an adjusted analytical result is required to take into consideration the results of interlaboratory trials. The analytical results obtained in accordance with clauses 7, 8 and 9 shall be adjusted by subtracting the analytical correction in table 2 to obtain an adjusted analytical result.

Materials are deemed to comply with the requirements of this part of ISO 8124, if the adjusted analytical result for the migrated element is less than or equal to the value given in table 1.

1) To be published.

Table 1 — Maximum acceptable element migration from toy materials

Values in milligrams per kilogram toy material

Toy material	Element							
	Sb	As	Ba	Cd	Cr	Pb	Hg	Se
Any toy material given in clause 1, except modelling clay and finger paint	60	25	1 000	75	60	90	60	500
Modelling clay and finger paint	60	25	250	50	25	90	25	500

Table 2 — Analytical correction

Element	Sb	As	Ba	Cd	Cr	Pb	Hg	Se
Analytical correction (%)	60	60	30	30	30	30	50	60

EXAMPLE:

An analytical result for lead of 120 mg/kg was obtained. The necessary analytical correction taken from table 2 is 30 %. Therefore the adjusted analytical result is

$$120 - \frac{120 \times 30}{100} = 120 - 36$$

$$= 84 \text{ mg/kg.}$$

This is deemed as complying with the requirements of this part of ISO 8124 (maximum acceptable migration of lead as given in table 1 is 90 mg/kg).

5 Principle

Soluble elements are extracted from toy materials under conditions which simulate the material remaining in contact with stomach acid for a period of time after swallowing. The concentrations of the soluble elements are then determined quantitatively by unspecified analytical methods with specified detection limits.

6 Reagents and apparatus

NOTE — No recommendation is made for the reagents, materials, and apparatus necessary for carrying out elemental analyses within the detection limits specified in clause 9.

6.1 Reagents

During the analyses, use only reagents of recognized analytical grade (see annex D).

6.1.1 Hydrochloric acid solution, $c(\text{HCl}) = (0,07 \pm 0,005) \text{ mol/l}$.

6.1.2 Hydrochloric acid solution, $c(\text{HCl}) = (0,14 \pm 0,010) \text{ mol/l}$.

6.1.3 Hydrochloric acid solution, $c(\text{HCl}) =$ approximately 1 mol/l.

6.1.4 Hydrochloric acid solution, $c(\text{HCl}) =$ approximately 2 mol/l.

6.1.5 Hydrochloric acid solution, $c(\text{HCl}) =$ approximately 6 mol/l.

6.1.6 1,1,1-trichloroethane, containing a maximum of 10 mg/kg of hydrochloric acid when tested in accordance with annex A, or other suitable solvent (see annex D).

6.1.7 Water of at least grade 3 purity, in accordance with ISO 3696.

6.2 Apparatus

Normal laboratory apparatus and

6.2.1 Plain-weave wire-cloth stainless steel metal sieve, of nominal aperture 0,5 mm and tolerances as indicated in table B.1.

6.2.2 Means of measuring pH with an accuracy of $\pm 0,2$ pH units. Cross-contamination shall be prevented (see annex D).

6.2.3 Membrane filter, of pore size 0,45 μm .

6.2.4 Centrifuge, capable of centrifuging at $(5\,000 \pm 500)g^{1)}$ (see annex D).

6.2.5 Means to agitate the mixture at a temperature of (37 ± 2) °C.

6.2.6 Series of containers, of gross volume between 1,6 times and 5,0 times that of the volume of hydrochloric acid extractant (see annex D).

7 Selection of test portions

A laboratory sample for testing shall consist of a toy either in the form in which it is marketed, or in the form in which it is intended to be marketed. Test portions shall be taken from accessible parts (see ISO 8124-1) of a single toy sample. When appropriate, the toy shall be subjected to relevant tests in accordance with ISO 8124-1, before the accessibility is considered. Identical materials in the toy may be combined and treated as a single test portion, but additional toy samples shall not be used. Test portions may be composed of more than one material or colour only if physical separation, e.g. dot printing, patterned textiles or mass limitation reasons, precludes the formation of discrete specimens. (See annex D.)

NOTE — The requirement does not preclude the taking of reference portions from toy materials in a different form provided that they are representative of the relevant material specified above and the substrate upon which they are deposited. (See annex D.)

Test portions of less than 10 mg of material shall not be tested.

1) $g = 9,806\,65\text{ m/s}^2$

8 Preparation and extraction of test portions

8.1 Coatings of paint, varnish, lacquer, printing ink, polymer and similar coatings

8.1.1 Test portion preparation

Remove the coating from the laboratory sample by scraping (3.5) at room temperature and comminute it at a temperature not exceeding ambient. Collect enough coating to obtain a test portion of preferably not less than 100 mg which will pass through a metal sieve of aperture 0,5 mm (6.2.1).

If only between 10 mg and 100 mg of comminuted uniform coating is available, extract this in accordance with 8.1.2 and calculate the quantity of the appropriate elements as if a test portion of 100 mg had been used. Report the mass of the test portion under 10 e).

In the case of coatings that by their nature cannot be comminuted (e.g. elastic/plastic paint), remove a test portion of coating from the laboratory sample without comminuting.

8.1.2 Extraction procedure

Using a container of appropriate size (6.2.6), mix the test portion prepared in 8.1.1 with 50 times its mass of an aqueous HCl solution at (37 ± 2) °C of $c(\text{HCl})$ 0,07 mol/l (6.1.1). [Where the test portion has only a mass of between 10 mg and 100 mg, mix the test portion with 5,0 ml of this solution (6.1.1) at (37 ± 2) °C.]

Shake for 1 min. Check the acidity of the mixture. If the pH is greater than 1,5, add dropwise, while shaking the mixture, an aqueous solution of $c(\text{HCl})$ approximately 2 mol/l (6.1.4) until the pH of the mixture is between 1,0 and 1,5.

Protect the mixture from light. Agitate the mixture continuously at (37 ± 2) °C (6.2.5) for 1 h and then allow to stand for 1 h at (37 ± 2) °C.

Without delay, efficiently separate the solids from the solution, firstly by filtration using a membrane filter (6.2.3) and, if necessary, by centrifuging at up to 5 000g (6.2.4). Carry out the separation as rapidly as possible after completion of the standing time. If centrifuging is used, it shall take no longer than 10 min and shall be reported under 10 e).

If the resulting solutions are to be stored for more than one working day prior to elemental analysis, stabilize them by addition of hydrochloric acid so that the concentration of the stored solution is approximately $c(\text{HCl}) = 1$ mol/l. Report such stabilization under 10 e).

8.2 Polymeric and similar materials, including laminates, whether textile-reinforced or not, but excluding other textiles

8.2.1 Test portion preparation

Obtain a test portion of preferably not less than 100 mg of the polymeric or similar materials, whilst avoiding heating of the materials, according to the following procedure.

Cut out test portions from those areas having the thinnest material cross-section in order to ensure a surface area of the test pieces as large as possible in proportion to their mass. Each piece shall in the uncompressed condition have no dimension greater than 6 mm.

If the laboratory sample is not of a uniform material, obtain a test portion from each different material present in a mass greater than 10 mg. Where there is only between 10 mg and 100 mg of uniform material, report the mass of the test portion under 10 e) and calculate the quantity of the appropriate elements as if a test portion of 100 mg had been used.

8.2.2 Extraction procedure

Follow the extraction procedure in 8.1.2 using the test portions prepared in accordance with 8.2.1.

8.3 Paper and paper board

8.3.1 Test portion preparation (see annex D)

Obtain a test portion of preferably not less than 100 mg of the paper or paper board.

If the laboratory sample is not of a uniform material, obtain a test portion from each different material present in a mass of not less than 100 mg. Where there is only between 10 mg and 100 mg of uniform material, report the mass of the test portion under 10 e) and calculate the quantity of the appropriate elements as if a test portion of 100 mg had been used.

If the paper or paper board to be tested is coated with paint, varnish, lacquer, printing ink, adhesive or similar coating, test portions of the coating shall not be taken separately. In such cases, take test portions from the material so that they also include representative parts of the coated area and report this under 10 e). Extract test portions so obtained in accordance with 8.3.2. (See annex D.)

8.3.2 Extraction procedure

Macerate the test portion prepared in 8.3.1 with 25 times its mass of water (6.1.7) at $(37 \pm 2)^\circ\text{C}$ so that the resulting mixture is homogeneous. Quantitatively transfer the mixture to the appropriate-sized container (6.2.6). Add to the mixture a mass of aqueous solution of $c(\text{HCl}) = 0,14 \text{ mol/l}$ (6.1.2) at $(37 \pm 2)^\circ\text{C}$ which has 25 times the mass of the test portion.

Shake for 1 min. Check the acidity of the mixture. If the pH is greater than 1,5, add dropwise, while shaking the mixture, an aqueous solution of $c(\text{HCl})$ approximately 2 mol/l (6.1.4) until the pH of the mixture is between 1,0 and 1,5.

Protect the mixture from light. Agitate the mixture continuously at $(37 \pm 2)^\circ\text{C}$ (see 6.2.5) for 1 h and then allow to stand for 1 h at $(37 \pm 2)^\circ\text{C}$.

Without delay, efficiently separate the solids from the solution, firstly by filtration using a membrane filter (6.2.3) and, if necessary, by centrifuging at up to 5 000g (see 6.2.4). Carry out the separation as rapidly as possible after completion of the standing time. If centrifuging is used, it shall take no longer than 10 min and shall be reported under 10 e).

If the resulting solutions are to be stored for more than one working day prior to elemental analysis, stabilize them by addition of hydrochloric acid so that the concentration of the stored solution is approximately $c(\text{HCl}) = 1 \text{ mol/l}$. Report such stabilization under 10 e).

8.4 Natural or synthetic textiles

8.4.1 Test portion preparation

Obtain a test portion of preferably not less than 100 mg by cutting the textile material into pieces which in the uncompressed condition have no dimension greater than 6 mm. (See annex D.)

If the sample is not of a uniform material or colour, obtain a test portion from each different material or colour present in a mass greater than 100 mg. Materials or colours present in amounts between 10 mg and 100 mg shall form part of the test portion obtained from the main material.

Samples taken from patterned textiles shall be representative of the whole material. (See annex D.)

8.4.2 Extraction procedure

Follow the extraction procedure in 8.1.2 using the test portions prepared in accordance with 8.4.1.

8.5 Glass/ceramic/metallic materials

8.5.1 Test portion preparation

Toys and toy components shall be first subjected to the small parts test in accordance with ISO 8124-1. If the toy or component fits entirely within the small parts cylinder and contains accessible glass, ceramic or metallic materials, then the toy or component shall be extracted in accordance with 8.5.2 after removal of any coating in accordance with 8.1.1. (See annex D.)

NOTE — Toys and toy components that have no accessible glass, ceramic or metallic materials do not require extraction in accordance with 8.5.2. (See annex D.)

8.5.2 Extraction procedure

Place the toy or toy component in a 50-ml glass cylinder with a nominal height of 60 mm and diameter of 40 mm.

NOTE — This type of container will take all components/toys that fit inside the small parts cylinder defined in ISO 8124-1.

Add a sufficient volume of an aqueous solution of $c(\text{HCl}) = 0,07 \text{ mol/l}$ (6.1.1) at $(37 \pm 2) \text{ }^\circ\text{C}$ to just cover the toy or component. Cover the container, protect the contents from light and allow the contents to stand for 2 h at $(37 \pm 2) \text{ }^\circ\text{C}$.

Without delay, efficiently separate the solids from the solution, firstly by decantation followed by filtration using a membrane filter (6.2.3) and, if necessary, by centrifuging at up to $5\,000g$ (6.2.4). Carry out the separation as rapidly as possible after completion of the standing time. If centrifuging is used, it shall take no longer than 10 min and shall be reported under 10 e).

If the resulting solutions are to be stored for more than one working day prior to elemental analysis, stabilize them by addition of hydrochloric acid so that the concentration of the stored solution is approximately $c(\text{HCl}) = 1 \text{ mol/l}$. Report such stabilization under 10 e).

8.6 Other materials, whether mass-coloured or not (see annex D)

8.6.1 Test portion preparation

Obtain a test portion of preferably not less than 100 mg of the material in accordance with 8.2.1, 8.3.1, 8.4.1 or 8.5.1, as appropriate.

If the laboratory sample is not of uniform material, a test portion shall be obtained from each different material present in a mass greater than 10 mg. Where there is only between 10 mg and 100 mg of uniform material, report the mass of the test portion under 10 e) and calculate the quantity of the appropriate elements as if a test portion of 100 mg had been used.

If the material to be tested is coated with paint, varnish, lacquer, printing ink or similar coating, follow the procedure in 8.1.1.

8.6.2 Extraction procedures

Extract the materials in accordance with 8.2.2, 8.3.2, 8.4.2 or 8.5.2, as appropriate. Report the method used under 10 e).

8.7 Materials intended to leave a trace

8.7.1 Test portion preparation for materials in solid form

Obtain a test portion of preferably not less than 100 mg by cutting the material into pieces which in the uncompressed condition have no dimension greater than 6 mm.

A test portion shall be obtained from each different material intended to leave a trace, present in the laboratory sample in mass greater than 10 mg. Where there is only between 10 mg and 100 mg of material, report the mass of the test portion under 10 e) and calculate the quantity of the appropriate elements as if a test portion of 100 mg had been used.

If the material contains any grease, oil, wax or similar material, enclose the test portion in hardened filter paper and remove these ingredients with 1,1,1-trichloroethane or other suitable solvent (6.1.6) by extraction before treatment of the test portion as described in 8.7.4. Take analytical measures to ensure that removal of the ingredients referred to is quantitative. Report the solvent used under 10 e).

8.7.2 Test portion preparation for materials in liquid form

Obtain a test portion of preferably not less than 100 mg of the material from the laboratory sample. The use of an appropriate solvent to facilitate the obtaining of a test portion is permitted.

A test portion shall be obtained from each different material intended to leave a trace, present in the laboratory sample in mass greater than 10 mg. Where there is only between 10 mg and 100 mg of material, report the mass of the test portion under 10 e) and calculate the quantity of the appropriate elements as if a test portion of 100 mg had been used.

If the material is intended to solidify in normal use and contains grease, oil, wax or similar material, allow the test portion to solidify under normal-use conditions and enclose the resulting material in hardened filter paper. Remove the grease, oil, wax or similar material with 1,1,1-trichloroethane or other suitable solvent (6.1.6) by extraction before treatment of the test portion as described in 8.7.4. Take analytical measures to ensure that removal of the ingredients referred to is quantitative. Report the solvent used under 10 e).

8.7.3 Extraction procedure for samples not containing grease, oil, wax or similar material

Using a container of appropriate size (6.2.6), mix the test portion prepared in accordance with 8.7.1 or 8.7.2 with 50 times its mass of an aqueous HCl solution at $(37 \pm 2)^\circ\text{C}$ of $c(\text{HCl}) = 0,07 \text{ mol/l}$ (6.1.1). For a test portion of mass between 10 mg and 100 mg, mix the test portion with 5,0 ml of this solution at $(37 \pm 2)^\circ\text{C}$.

Shake for 1 min. Check the acidity of the mixture. If the test portion contains large quantities of alkaline materials, generally in the form of calcium carbonate, adjust the pH to between 1,0 and 1,5 with hydrochloric acid [$c(\text{HCl})$ approximately 6 mol/l (6.1.5)] in order to avoid overdilution. Report the amount of hydrochloric acid used to adjust pH in relation to the total amount of solution under 10 e).

If only a small quantity of alkaline material is present and the pH of the mixture is greater than 1,5, add dropwise, while shaking the mixture, an aqueous solution of $c(\text{HCl})$ approximately 2 mol/l (6.1.4) until the pH is between 1,0 and 1,5.

Protect the mixture from light. Agitate the mixture continuously at $(37 \pm 2)^\circ\text{C}$ (6.2.5) for 1 h and then allow to stand for 1 h at $(37 \pm 2)^\circ\text{C}$ prior to elemental analysis.

8.7.4 Extraction procedure for samples containing grease, oil, wax or similar material

With the test portion as prepared in 8.7.1 or 8.7.2 remaining in the hardened filter paper, macerate the test portion with a mass of water (6.1.7) at $(37 \pm 2)^\circ\text{C}$ which has 25 times the mass of the original material so that the resulting mixture is homogeneous. Quantitatively transfer the mixture to a container of appropriate size (6.2.6). Add to the mixture a mass of aqueous solution of $c(\text{HCl}) = 0,14 \text{ mol/l}$ (6.1.2) at $(37 \pm 2)^\circ\text{C}$ which has 25 times the mass of the original test portion.

In the case of a test portion of original mass between 10 mg and 100 mg, macerate the test portion with 2,5 ml or water (6.1.7). Quantitatively transfer the mixture to the appropriate-sized container (6.2.6). Add 2,5 ml aqueous solution of $c(\text{HCl}) = 0,14 \text{ mol/l}$ (6.1.2) at $(37 \pm 2)^\circ\text{C}$ to the mixture.

Shake for 1 min. Check the acidity of the mixture. If the test portion contains large quantities of alkaline materials, generally in the form of calcium carbonate, adjust the pH to between 1,0 and 1,5 with hydrochloric acid of $c(\text{HCl})$ approximately 6 mol/l (6.1.5) in order to avoid overdilution. Report the amount of hydrochloric acid used to adjust pH in relation to the total amount of solution under 10 e).

If only a small quantity of alkaline material is present and the pH of the mixture is greater than 1,5, add dropwise, while shaking the mixture, an aqueous solution of $c(\text{HCl})$ approximately 2 mol/l (6.1.4) until the pH is between 1,0 and 1,5.

Protect the mixture from light. Agitate the mixture continuously at $(37 \pm 2) ^\circ\text{C}$ (6.2.5) for 1 h and then allow to stand for 1 h at $(37 \pm 2) ^\circ\text{C}$.

NOTE — The volume of HCl solution of $c(\text{HCl}) = 0,07 \text{ mol/l}$ (see 8.7.3) or $c(\text{HCl}) = 0,14 \text{ mol/l}$, as the case may be, is calculated on the mass of the test portion prior to dewaxing.

Without delay, efficiently separate the solids from the solution, firstly by filtration using a membrane filter (6.2.3) and, if necessary, by centrifuging at up to 5 000g (6.2.4). Carry out the separation as rapidly as possible after completion of the standing time. If centrifuging is used, it shall take no longer than 10 min and shall be reported under 10 e).

If the resulting solutions are to be stored for more than one working day prior to elemental analysis, stabilize them by addition of hydrochloric acid so that the concentration of the stored solution is approximately $c(\text{HCl}) = 1 \text{ mol/l}$. Report such stabilization under 10 e).

8.8 Pliable modelling materials, including modelling clays, and gels

8.8.1 Test portion preparation

Obtain a test portion of not less than 100 mg of the material from the laboratory sample; obtain a test portion from each different material in the laboratory sample.

If the material contains grease, oil, wax or similar material, enclose the test portion in hardened filter paper and remove these ingredients with 1,1,1-trichloroethane or other suitable solvent (6.1.6) by extraction before treatment of the test portion as described in 8.8.3. Take analytical measures to ensure that removal of the ingredients referred to is quantitative. Report solvent used under 10 e).

8.8.2 Extraction procedure for samples not containing grease, oil, wax or similar material

Using a container of appropriate size (6.2.6) mix the test portion prepared in accordance with 8.8.1, after breaking up of clay or doughy materials if appropriate, with 50 times its mass of an aqueous solution at $(37 \pm 2) ^\circ\text{C}$ of $c(\text{HCl}) = 0,07 \text{ mol/l}$ (6.1.1).

Shake the mixture for 1 min. Check the acidity of the mixture. If the test portion contains large quantities of alkaline materials, generally in the form of calcium carbonate, adjust the pH to between 1,0 and 1,5 with hydrochloric acid of $c(\text{HCl})$ approximately 6 mol/l (6.1.5) in order to avoid overdilution. Report the amount of hydrochloric acid used in relation to the total amount of solution under 10 e).

If only a small quantity of alkaline material is present and the pH of the mixture is greater than 1,5, add dropwise, while shaking the mixture, an aqueous solution of $c(\text{HCl})$ approximately 2,0 mol/l (6.1.4) until the pH is between 1,0 and 1,5.

Protect the mixture from light. Agitate the mixture continuously at $(37 \pm 2) ^\circ\text{C}$ (6.2.5) for 1 h and then allow to stand for 1 h at $(37 \pm 2) ^\circ\text{C}$.

8.8.3 Extraction procedure for samples containing grease, oil, wax or similar material

With the test portion as prepared in 8.8.1 remaining in the hardened filter paper, macerate the test portion with a mass of water (6.1.7) at $(37 \pm 2) ^\circ\text{C}$ which has 25 times the mass of the original material so that the resulting mixture is homogeneous. Quantitatively transfer the mixture to a container of appropriate size (6.2.6). Add to the mixture a mass of aqueous solution of $c(\text{HCl}) = 0,14 \text{ mol/l}$ (6.1.2) at $(37 \pm 2) ^\circ\text{C}$ which has 25 times the mass of the original test portion.

Shake for 1 min. Check the acidity of the mixture. If the test portion contains large quantities of alkaline materials, generally in the form of calcium carbonate, adjust the pH to between 1,0 and 1,5 with an aqueous solution of $c(\text{HCl})$ approximately 6 mol/l (6.1.5) in order to avoid overdilution. Report the amount of hydrochloric acid used in relation to the total amount of solution under 10 e).

If only a small quantity of alkaline material is present and the pH of the mixture is greater than 1,5, add dropwise, while shaking the mixture, an aqueous solution of $c(\text{HCl})$ approximately 2,0 mol/l (6.1.4) until the pH is between 1,0 and 1,5.

Protect the mixture from light. Agitate the mixture continuously at $(37 \pm 2) ^\circ\text{C}$ (6.2.5) for 1 h and then allow to stand for 1 h at $(37 \pm 2) ^\circ\text{C}$.

NOTE — The volume of the HCl solution of $c(\text{HCl}) = 0,07 \text{ mol/l}$ (see 8.8.2) or $c(\text{HCl}) = 0,14 \text{ mol/l}$, as the case may be, is calculated on the mass of the test portion prior to dewaxing.

Without delay, efficiently separate the solids from the solution, firstly by filtration using a membrane filter (6.2.3) and, if necessary, by centrifuging at up to 5 000g (6.2.4). Carry out the separation as rapidly as possible after completion of the standing time. If centrifuging is used, it shall take no longer than 10 min and shall be reported under 10 e).

If the resulting solutions are to be stored for more than one working day prior to elemental analysis, stabilize them by addition of hydrochloric acid so that the concentration of the stored solution is approximately $c(\text{HCl}) = 1 \text{ mol/l}$. Report such stabilization under 10 e).

8.9 Paints, including finger paints, varnishes, lacquers, glazing powders and similar materials in solid or liquid form

8.9.1 Test portion preparation for materials in solid form

Obtain a test portion of preferably not less than 100 mg of the material, if appropriate by scraping off the material or by cutting into pieces which in the uncompressed condition have no dimension greater than 6 mm.

A test portion shall be obtained from each different material present in the laboratory sample in mass greater than 10 mg. Where there is only between 10 mg and 100 mg of material, report the mass of the test portion under 10 e) and calculate the quantity of the appropriate elements as if a test portion of 100 mg had been used.

If the material contains any grease, oil, wax or similar material, enclose the test portion in hardened filter paper and remove these ingredients with 1,1,1-trichloroethane or other suitable solvent (6.1.6) by extraction before treatment of the test portion as described in 8.9.4. Take analytical measures to ensure that removal of the ingredients referred to is quantitative. Report the solvent used under 10 e).

If the test portion is removed by scraping, comminute the test portion so that the material will pass through a metal sieve with an aperture of 0,5 mm (6.2.1).

8.9.2 Test portion preparation for materials in liquid form

Obtain a test portion of preferably not less than 100 mg of the material from the laboratory sample. The use of an appropriate solvent to facilitate the obtaining of a test portion is permitted.

A test portion shall be obtained from each different material present in the laboratory sample and forming a mass greater than 10 mg. Where there is only between 10 mg and 100 mg of material, report the mass of the test portion under 10 e) and calculate the quantity of the appropriate elements as if a test portion of 100 mg had been used.

If the material is intended to solidify in normal use and contains grease, oil, wax or similar material, allow the test portion to solidify under normal-use conditions and enclose the resulting material in hardened filter paper. Remove the grease, oil, wax or similar material with 1,1,1-trichloroethane or other suitable solvent (6.1.6) by extraction before treatment of the test portion as described in 8.9.4. Take analytical measures to ensure that removal of the ingredients referred to is quantitative. Report the solvent used under 10 e).

8.9.3 Extraction procedure for samples not containing grease, oil, wax or similar material

Follow the extraction procedure in 8.7.3 using the test portions prepared in accordance with 8.9.1 or 8.9.2.

8.9.4 Extraction procedure for samples containing grease, oil, wax or similar material

With the test portion as prepared in 8.9.1 or 8.9.2 remaining in the filter paper, macerate the test portion with a mass of water (6.1.7) at $(37 \pm 2)^\circ\text{C}$ which has 25 times the mass of the original material so that the resulting mixture is homogeneous. Quantitatively transfer the mixture to a container of appropriate size (6.2.6). Add to the mixture a mass of aqueous solution of $c(\text{HCl}) = 0,14 \text{ mol/l}$ (6.1.2) at $(37 \pm 2)^\circ\text{C}$ which has 25 times the mass of the original test portion.

Shake for 1 min. Check the acidity of the mixture. If the test portion contains large quantities of alkaline materials, generally in the form of calcium carbonate, adjust the pH to between 1,0 and 1,5 with an aqueous solution of $c(\text{HCl})$ approximately 6 mol/l (6.1.5) in order to avoid overdilution. Report the amount of hydrochloric acid used in relation to the total amount of solution under 10 e).

If only a small quantity of alkaline material is present and the pH of the mixture is greater than 1,5, add dropwise, while shaking the mixture, an aqueous solution of $c(\text{HCl})$ approximately 2 mol/l (6.1.4) until the pH is between 1,0 and 1,5.

Protect the mixture from light. Agitate the mixture continuously at $(37 \pm 2)^\circ\text{C}$ (6.2.5) for 1 h and then allow to stand for 1 h at $(37 \pm 2)^\circ\text{C}$.

NOTE — The volume of the solution of $c(\text{HCl}) = 0,07 \text{ mol/l}$ (see 8.7.3) or $c(\text{HCl}) = 0,14 \text{ mol/l}$, as the case may be, is calculated on the mass of the test portion prior to dewaxing.

Without delay, efficiently separate the solids from the solution, firstly by filtration using a membrane filter (6.2.3) and, if necessary, by centrifuging at up to 5 000g (6.2.4). Carry out the separation as rapidly as possible after completion of the standing time. If centrifuging is used, it shall take no longer than 10 min and shall be reported under 10 e).

If the resulting solutions are to be stored for more than one working day prior to elemental analysis, stabilize them by addition of hydrochloric acid so that the concentration of the stored solution is approximately $c(\text{HCl}) = 1 \text{ mol/l}$. Report such stabilization under 10 e).

9 Detection limits of quantitative elemental analysis

For the quantitative analysis of the extracts of toys for the migrated elements listed in clause 1, methods having a detection limit of a maximum of 1/10 of the values to be determined (see 4.1, table 1) shall be applied. The detection limit (of the analytical method) is deemed to be three times the standard deviation of the blank value as measured by the laboratory carrying out the analysis of the toy materials.

Laboratories using methods deviating from this requirement shall report the detection limit under 10 c).

10 Test report

The test report shall contain at least the following information:

- type and identification of the product and/or material tested;
- a reference to this part of ISO 8124 (ISO 8124-3:1997);
- the technique used for determining the quantity of each migrated element, and the detection limit if different from the requirements in clause 9;

- d) the adjusted results (see 4.2) of the quantitative elemental analysis, expressed as milligrams migrated element per kilogram toy material, stating that the result is related to the element in solution;
- e) details of the procedure used, in accordance with requirements in clause 8;
- f) any departure, by agreement or otherwise, from the preparation and extraction procedures specified;
- g) date of the test.

Annex A (normative)

Determination of acidity of 1,1,1-trichloroethane

A.1 Reagents

A.1.1 Standard sodium hydroxide solution, $c(\text{NaOH}) = (0,1 \pm 0,005) \text{ mol/l}$.

A.1.2 Phenolphthalein indicator, 0,5 g in 100 ml of 95 % (V/V) (volume fraction 95 %) ethanol, turned faintly pink by the addition of the minimum quantity of dilute aqueous hydrochloric acid or dilute aqueous sodium hydroxide.

A.1.3 Water, neutralized by the addition of dilute aqueous sodium hydroxide using a few drops of the phenolphthalein indicator.

A.2 Test procedure

Place 100 ml of neutralized water (A.1.3) in a 250 ml flask, add 100 ml of 1,1,1-trichloroethane test sample, stopper the flask and shake vigorously. Allow the layers to separate, add 0,5 cm³ of phenolphthalein indicator and titrate with 0,1 mol/l aqueous sodium hydroxide (A.1.1) using a microburette until the upper layer turns a faint pink colour.

A.3 Calculation

The acidity, A , of 1,1,1-trichloroethane, expressed as micrograms of hydrochloric acid per gram of trichloroethane, is calculated from the following equation:

$$A = \frac{36,5}{\rho} V$$

where

V is the volume of 0,1 mol/l sodium hydroxide, expressed in millilitres;

ρ is the relative density of the test sample of 1,1,1-trichloroethane.

Annex B
 (normative)

Sieve requirements

Table B.1 — Sieve dimensions and tolerances

Dimensions in millimetres

Nominal aperture size	Nominal wire diameter in test sieve	Tolerances		
		Maximum deviation for size of an individual aperture	Tolerance for average aperture	Intermediate deviation (no more than 6 % of the apertures shall exceed the nominal aperture size plus this number)
0,500	0,315	+ 0,090	± 0,018	+ 0,054

Annex C (informative)

Selection of procedure

The diagram in figure C.1 is a guideline for the choice of procedure to be used for the various toy materials.

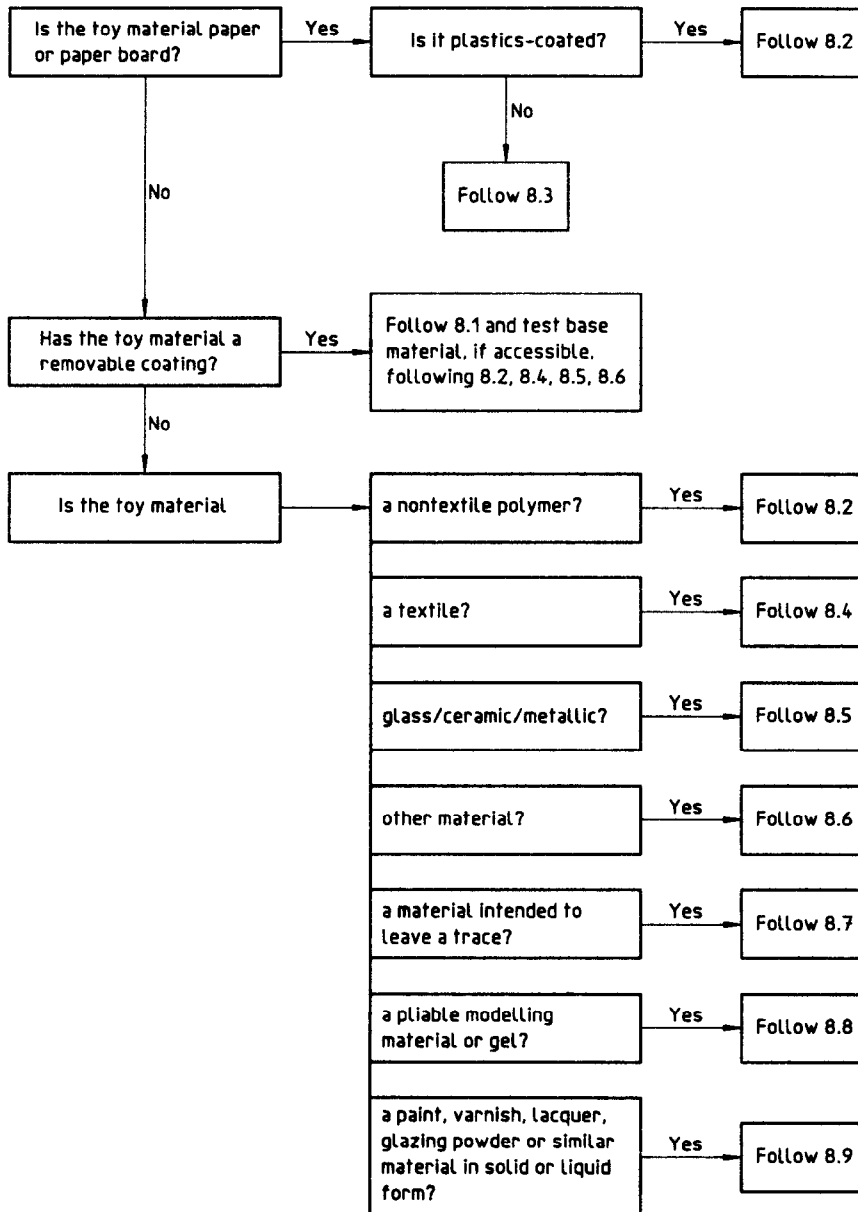


Figure C.1 — Guidelines for choosing the procedure for preparation and extraction of test portions

Annex D (informative)

Background and rationale

D.1 Introduction

The way bioavailability is defined in the European Council Directive 88/378/EEC of May 1988 concerning the safety of toys (published in the Official Journal of the EC No. L 187 of 16 July 1988) led to the methods in this part of ISO 8124 addressing the migration levels of soluble elements from a toy material.

The approach using total migrated element determinations was discarded for the following reasons.

- a) The above-mentioned Directive indicates bioavailability limits, and there has been no link to date between the availability of an element in a toy material with respect to extraction with simulated gastric solutions and the total content of that element in the material;
- b) Some compounds, such as barium sulfate, can be included in products at a 2 % level to render them radio-opaque. This use of barium would either have to be excluded from toys or a separate requirement made.
- c) Cadmium compounds can be used as stabilizers in plastics such as polyvinylchloride (PVC). These compounds under normal conditions are insoluble in simulated gastric solution. Their use therefore, if total elements were determined, would have to be accounted for in the requirements of this part of ISO 8124. A case can be made for selenium, which can be present as a constituent of insoluble pigments, etc. (See also D.4).

D.2 Scope

D.2.1 Subclause 1.3

Subclause 1.3 is intended to indicate an approach to the decision of what toys, or toy components, are excluded from this part of ISO 8124 because of characteristics that render them unlikely to present a risk of injury from the ingestion of materials containing toxic elements.

This was considered a logical approach for a number of reasons, including:

- as children get older, the tendency for them to place toys/toy materials in their mouths is reduced and therefore the risk of the ingestion of toxic elements is reduced;
- the larger the toy or the less accessible the material, the lower the risk of ingestion of toxic elements.

It was therefore considered that all toys which might be placed in the mouth, or close to the mouth would be tested, e.g. pencils.

Toys intended for children over the age of six years were considered not to pose a significant risk of injury through the ingestion of toxic elements.

D.2.2 Packaging (subclause 1.4)

“Unless part of the toy” in 1.4 is intended to mean, for example, boxes containing jigsaw puzzles or packaging on which the instructions are included in the case of games, etc., but taking into account the second listed item of 1.3, restricting the requirements to toys intended for children up to six years of age. It is not intended to address, for example, blister packs containing simple instructions.

D.3 Specific requirements (see 4.1)

The maximum acceptable level of soluble barium has been raised from 500 mg/kg to 1 000 mg/kg for the following reasons:

- the use of barium sulfate in toys has led to levels of soluble barium in the acidic extraction solution [$c(\text{HCl}) = (0,07 \pm 0,005) \text{ mol/l}$ at 37 °C] of between 400 mg/kg and 600 mg/kg as expressed on the mass of toy material taken. This level is such that, due to the statistical uncertainty of the determination, a PASS or FAIL could not be indicated;
- the formation of non-bioavailable colloidal barium sulfate crystals in the filtrate resulted in levels of apparently soluble barium which exceeded 500 mg/kg because of problems with filtration;
- in addition, the previous maximum level of migration of barium of 500 mg/kg from toy materials was not consistent with 25,0 µg/d bioavailability and the 8 mg/d of toy material intake; 25,0 µg corresponds to a migration limit of 3,125 mg/kg. It is understood that the 500 mg/kg limit was consciously selected despite the "theoretical" 3,125 mg/kg value. The effect of the 500 mg/kg limit was to reduce the bioavailability from the advised 25 µg to 4 µg. It should be noted that the 25,0 µg value has been reduced from an initial 50,0 µg, not for toxicological reasons, but according to the Commission of the European Communities, "to reduce the avoidable input on body burden".

D.4 Statistical uncertainty of the test procedure and interpretation of results (see 4.2)

Most chemical test methods are designed to measure the *total* amount of a substance in a material. This makes it easier to obtain an accurate result with close statistical agreement between laboratories because there is an absolute or true value.

Because of the way bioavailability is defined in the European Council Directive 88/378/EEC, the methods in this part of ISO 8124 lead to measurement of the migration of *soluble* element from a toy material. With this approach, the analytical result is dependent upon the specified conditions of test and there is no absolute or true value. Consequently, it is more difficult to obtain close statistical agreement between laboratories when performing such migration tests.

This is illustrated by the statistical information in EN 71-3:1988 taken from a 1987 European interlaboratory trial involving 17 laboratories. Results on an identical material varied by at least 30 % and up to 50 % between laboratories, depending on the instrumental technique used to measure the soluble element concentration of the filtrate. Moreover, these figures would be approximately three times higher if adjusted to a 95 % confidence level.

This degree of statistical uncertainty creates problems for manufacturers and enforcement authorities if test results are near the maximum limits allowed in this part of ISO 8124. Then it is not statistically possible to pass or fail a toy and leads to inconsistency in the interpretation of results.

There is no direct relationship between the actual total element content of a toy material and the soluble migration of that element under standard test conditions. Therefore, measuring the total element content and converting the result to give a soluble element value is not an answer to this problem. Setting maximum total element limits is a possibility, but requires an amendment to the European Council Directive 88/378/EEC (see also D.1).

Since 1988, the procedure for extraction from paint coatings on toys has been thoroughly investigated to determine which parameters significantly effect the results. The most critical parameters are the shape, size and mass of the paint particles produced by removal of the paint and its subsequent comminution. Other less critical parameters include the method of shaking, the temperature, and the type and porosity of filter paper.

As a result, a procedure for scraping and comminution to collect a paint test portion of particle size 300 µm to 500 µm was specified as a *proposed revision* to the procedure. A European interlaboratory trial was conducted in 1993 with 29 participating laboratories to compare the amended procedure with that in EN 71-3:1988.

The trial showed that results on an identical material can vary by at least 25 % and up to 80 % depending on the method of preparation of the test portions and the instrumental technique used to measure the soluble-element concentration of the filtrate.

Statistical agreement between laboratories showed improvement using a specified scraping procedure, but not when collecting a test portion of particle size 300 µm to 500 µm. However, any improvement was not significant enough to justify the proposed changes.

The trial confirmed that the use of different instrumental techniques contribute to the statistical uncertainty of the test procedure. It was also noted that laboratories need to check and calibrate their instruments on a regular basis to ensure accurate readings. Inductively Coupled Plasma (ICP) was more widely used by laboratories this time, and tended to show better agreement for most of the elements, particularly arsenic, antimony and selenium. However, it is not as sensitive as hydride-generation methods for low levels of the same elements.

A test procedure that produces results varying, at best, by 25 % between laboratories would normally be considered as technically unsuitable as a reference method. However, in reality, toys will either easily pass or fail this test, and only in relatively few cases will a result fall within the area of uncertainty. When this occurs, it is important that laboratories interpret the results in the same way.

It has been accepted that the test procedure cannot be improved without imposing time-consuming, costly, and in some cases, impractical burdens on laboratories with little benefit in terms of statistical agreement and safety. Therefore, the procedure allows laboratories to use their preferred techniques for scraping the paint off toys, collecting the portion that passes through a 500 µm sieve and determining the soluble-element concentration of the filtrate.

To achieve consistent interpretation of results, a correction factor for each element and applicable to all instrumental techniques has been introduced into this part of ISO 8124. These are taken from the precision data in EN 71-3:1988 and are used when an analytical result equals or exceeds the maximum limit. The analytical result is adjusted as described in 4.2, using the relevant correction factor. This way of interpreting the results is perfectly adequate as a screening test to differentiate between safe and unsafe toys as well as to ensure the safety of children.

In future, it is recommended that laboratories check and compare their performance of the test procedure by using reference materials and participating in a proficiency scheme.

D.5 Reagents (see 6.1)

D.5.1 1,1,1-trichloroethane (see 6.1.6)

This subclause limits the maximum level of acidity that 1,1,1-trichloroethane is permitted to contain. As is well known, this solvent breaks down in sunlight to form hydrochloric acid.

Under the Montreal Protocol, 1,1,1-trichloroethane will be phased out for general use. This part of ISO 8124 indicates that other suitable solvents may be used, but these will need to be evaluated to ensure that they are as efficient at wax/grease extraction as is 1,1,1-trichloroethane.

D.6 Apparatus (see 6.2)

D.6.1 Plain-weave wire cloth stainless steel sieve (see 6.2.1)

See D.4.

D.6.2 Means of measuring pH (see 6.2.2)

The measurement of pH is not restricted to the use of a pH-meter.

D.6.3 Centrifuge (see 6.2.4 and clause 8)

This subclause specifies the performance requirements of the centrifuge. Clause 8 specifies centrifuging limits and the amount of time permitted for centrifuging (to 10 min) and requires this to be reported under 10 e). The latter is necessary, as centrifuging has been reported to increase the extraction of barium.

D.6.4 Series of containers (see 6.2.6)

The indication of the gross volume of the containers is intended to ensure adequate movement of the solution, leading to a more efficient extraction.

D.7 Selection of test portions (see clause 7)

The practice of analysing "composite" (combination of different materials or colours) test portions is neither appropriate nor will it normally be necessary (with the availability of the "5,0 ml" test method). The analysis of composite materials is not satisfactory because, from a theoretical viewpoint, it can result in reduction of the migration of toxic elements that would otherwise occur. A simple example is known to have occurred, viz. barium extraction from paint was reduced when coextracted with another paint. This may have been the result of a counterion in the second paint which would cause the barium to be precipitated. Sulfate is such a counterion; whether sulfate itself was involved is not known and is not important for the establishment of this principle. Thus, except for cases where the separation of colours or toy material is impractical, e.g. dot printing, each discrete area is treated as a single sample.

The note makes it possible to test toy materials which are not in the form of a toy for *reference* purposes. However, this part of ISO 8124 clearly requires taking of *test* portions from the toy itself.

D.8 Paper and paper board — Test portion preparation (see 8.3.1)

Paper and paper board are to be treated as if they were a single material, i.e. surface coatings, if present, are not to be removed, but test portions must include *representative parts of the surface*. This procedure has been adopted because, in the practical situation of a child chewing paper or paper board, preferential removal of a coating is unlikely and the substrate is equally important.

D.9 Natural or synthetic textiles — Test portion preparation (see 8.4.1)

It is not feasible to take separate coloured test portions from a complex-patterned fabric. It is therefore required that a single test portion be taken that represents all the colours in the material.

D.10 Glass/ceramic/metallic materials — Test portion preparation (see 8.5.1)

Toys or components which do not fit entirely within the "small parts cylinder" of ISO 8124-1 are not tested because there is no hazard from ingestion and no significant extraction occurs with saliva simulator. The small parts cylinder is used to assess the size of toys/toy components for all relevant age groups. Comminution of glass, ceramic and metallic materials is inappropriate. Agitation of the test solution would be impractical for many examples and thus extraction is carried out without shaking. The diameter of the vessel and the orientation of the test portion have been selected to minimize variables.

Glass, ceramic and metallic materials completely coated so that no glass, ceramic or metal is accessible as defined in ISO 8124-1 are not tested in accordance with this requirement.

Where glass, ceramic and metal surfaces are accessible, even when partially covered by a coating, these are tested in accordance with 8.5.2 after complete removal of the partial coating in accordance with the method given in 8.1.1. This procedure is a compromise, as only a single toy may be taken as a sample as specified in clause 7 of this part of ISO 8124.

D.11 Other materials, whether mass-coloured or not (see 8.6)

This subclause applies to paper and paper board with a mass per unit area greater than 400 g/m² and may be fibreboard, hardboard, etc. In addition it applies to mass-coloured materials and other materials which are not coloured, such as wood, hardboard, leather and bone, etc. which may have received some other treatment, but were not covered by EN 71-3:1988.

Annex E (informative)

Bibliography

- [1] EN 71-3:1988, *Safety of toys — Migration of certain elements*.
- [2] European Council Directive 88/378/EEC of 3 May 1988 concerning the safety of toys (published in the Official Journal of the EC. No. L 187 of 16 July 1988).
- [3] European Council Directive 93/68/EEC of 22 July 1993, pp. 4 and 5, amending Directive 88/378/EEC concerning the safety of toys (published in the Official Journal of the EC No. L 220 of 30 August 1993).

Bureau of Indian Standards

BIS is a statutory institution established under the *Bureau of Indian Standards Act, 1986* to promote harmonious development of the activities of standardization, marking and quality certification of goods and attending to connected matters in the country.

Copyright

BIS has the copyright of all its publications. No part of these publications may be reproduced in any form without the prior permission in writing of BIS. This does not preclude the free use, in the course of implementing the standard, of necessary details, such as symbols and sizes, type or grade designations. Enquiries relating to copyright be addressed to the Director (Publications), BIS.

Review of Indian Standards

Amendments are issued to standards as the need arises on the basis of comments. Standards are also reviewed periodically; a standard along with amendments is reaffirmed when such review indicates that no changes are needed; if the review indicates that changes are needed, it is taken up for revision. Users of Indian Standards should ascertain that they are in possession of the latest amendments or edition by referring to the latest issue of 'BIS Handbook' and 'Standards : Monthly Additions'.

This Indian Standard has been developed from Doc : No. PCD 12 (1479).

Amendments Issued Since Publication

Amend No.	Date of Issue	Text Affected

BUREAU OF INDIAN STANDARDS

Headquarters:

Manak Bhavan, 9 Bahadur Shah Zafar Marg, New Delhi 110002
Telephones : 323 01 31, 323 94 02, 323 33 75

Telegrams: Manaksanstha
(Common to
all offices)

Regional Offices:

Central : Manak Bhavan, 9 Bahadur Shah Zafar Marg
NEW DELHI 110002

Telephone

{ 323 76 17
323 38 41

Eastern : 1/14 C. I. T. Scheme VII M, V. I. P. Road, Maniktola
CALCUTTA 700054

{ 337 84 99, 337 85 61
337 86 26, 337 86 62

Northern : SCO 335-336, Sector 34-A, CHANDIGARH 160022

{ 60 38 43
60 20 25

Southern : C. I. T Campus, IV Cross Road, CHENNAI 600113

{ 235 02 16, 235 04 42
235 15 19, 235 23 15

Western : Manakalaya, E9 MIDC, Marol, Andheri (East)
MUMBAI 400093

{ 832 92 95, 832 78 58
832 78 91, 832 78 92

Branches : AHMADABAD. BANGALORE. BHOPAL. BHUBANESHWAR.
COIMBATORE. FARIDABAD. GHAZIABAD. GUWAHATI. HYDERABAD. JAIPUR.
KANPUR. LUCKNOW. NAGPUR. PATNA. PUNE. THIRUVANANTHAPURAM.